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## Short Communications

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**Crystal structure of *n*-aliphatic acid hydrazides.** By L. H. JENSEN, *Department of Anatomy, School of Medicine, University of Washington, Seattle 5, Wash., U.S.A.* and E. C. LINGAFELTER, *Department of Chemistry and Chemical Engineering, University of Washington, Seattle 5, Wash., U.S.A.*

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In the course of investigating the structures of paraffin chain compounds, the *n*-aliphatic acid monosubstituted hydrazides have been studied.

Single crystals of hexanoic, heptanoic, octanoic, and nonanoic acid hydrazides were grown (these will be designated C<sub>6</sub>, C<sub>7</sub>, C<sub>8</sub> and C<sub>9</sub> hydrazides), and oscillation and Weissenberg photographs about the *a* and *b* axes were taken using Cu K $\alpha$  radiation. Preliminary dimensions of the monoclinic unit cells and densities are listed in Table 1.

The space group is either *Aa* or *A2/a*; with 8 mole-

cules in the unit cell, *A2/a* is probable. This was confirmed by applying to  $\{h0l\}$  of C<sub>7</sub> hydrazide the modification of Wilson's ratio test suggested by Howells, Phillips & Rogers (1950). The distribution of intensities follows fairly well that to be expected from a centrosymmetric projection (Fig. 1).

Intensities were estimated by eye for  $\{h0l\}$  of C<sub>7</sub> and

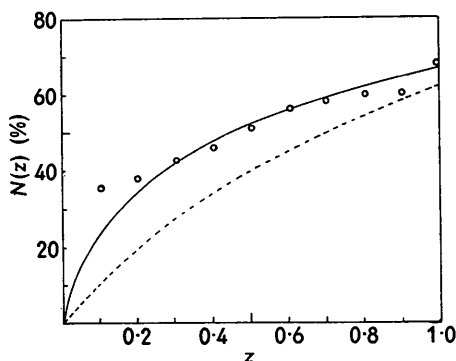


Fig. 1. Plot of  $N(z)$  v.  $z$  for  $\{h0l\}$  of heptanoic acid hydrazide. Solid line, centrosymmetric; broken line, non-centrosymmetric.

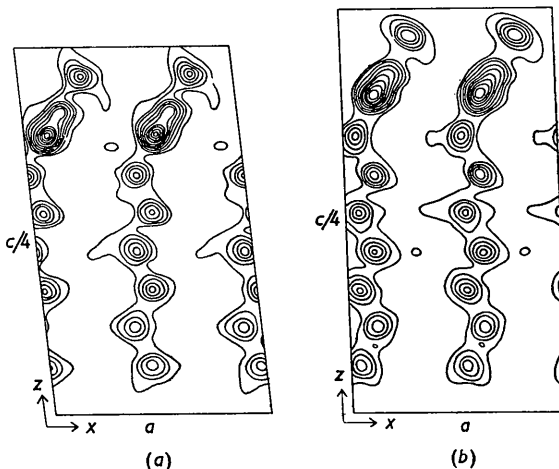


Fig. 2. Projection on (010) for (a) C<sub>7</sub> and (b) C<sub>8</sub> hydrazides. Contours at intervals of 1.5 e.Å<sup>-2</sup>. Origin may be as shown or at  $\frac{1}{2}$ , 0. Correct choice will be indicated by  $F_{hkl}$  or  $F_{hk0}$ .

Table 1. *Dimensions of unit cells and densities*

|   | <i>a</i> | <i>b</i> | <i>c</i> | $\beta$ | Density (g.cm. <sup>-3</sup> ) |        |
|---|----------|----------|----------|---------|--------------------------------|--------|
|   |          |          |          |         | Exp.                           | Calc.* |
| (C <sub>6</sub> )C <sub>5</sub> H <sub>11</sub> CONHNH <sub>2</sub> | 7.49 Å   | 4.88 Å   | 43.47 Å  | 91° 16' | 1.12                           | 1.09   |
| (C <sub>7</sub> )C <sub>6</sub> H <sub>13</sub> CONHNH <sub>2</sub> | 7.44     | 4.88     | 48.60    | 96 14   | 1.09                           | 1.09   |
| (C <sub>8</sub> )C <sub>7</sub> H <sub>15</sub> CONHNH <sub>2</sub> | 7.46     | 4.87     | 53.47    | 91 20   | 1.08                           | 1.08   |
| (C <sub>9</sub> )C <sub>8</sub> H <sub>17</sub> CONHNH <sub>2</sub> | 7.44     | 4.87     | 58.73    | 95 4    | 1.07                           | 1.08   |

\* Assuming 8 molecules per unit cell.

Table 2. Coordinates for C<sub>7</sub>, C<sub>8</sub> and C<sub>9</sub> hydrazides

|                | C <sub>7</sub> |       | C <sub>8</sub> |        | C <sub>9</sub> |
|----------------|----------------|-------|----------------|--------|----------------|
|                | x (Å)          | z (Å) | x (Å)          | z (Å)  | y (Å)          |
| C <sub>1</sub> | 3.56           | 1.57  | 0.33           | 1.57   | 1.57           |
| C <sub>2</sub> | 3.13           | 2.85  | 0.94           | 2.87   | 0.96           |
| C <sub>3</sub> | 3.88           | 4.11  | 0.39           | 4.12   | 1.48           |
| C <sub>4</sub> | 3.46           | 5.38  | 0.97           | 5.37   | 0.98           |
| C <sub>5</sub> | 4.21           | 6.65  | 0.44           | 6.64   | 1.67           |
| C <sub>6</sub> | 3.78           | 7.91  | 1.03           | 7.92   | 0.92           |
| C <sub>7</sub> | 4.28*          | 9.25* | 0.44           | 9.20   | 1.60           |
| C <sub>8</sub> | —              | —     | 1.13*          | 10.60* | 0.95           |
| C <sub>9</sub> | —              | —     | —              | —      | 1.99*          |
| O              | 4.28*          | 9.25* | 1.13*          | 10.60* | 1.99*          |
| N <sub>1</sub> | 5.16           | 10.00 | 1.66*          | 11.29* | 0.91           |
| N <sub>2</sub> | 5.97           | 11.18 | 2.41           | 12.48  | 1.24           |

\* Not resolved in *F* projection.

C<sub>8</sub> hydrazides and {0*kl*} for C<sub>9</sub> hydrazide. The  $\Delta d_{001}$  values indicate that the chains are very nearly normal to (001). Bragg-Lipson charts were used to determine the approximate structure. Electron densities were computed at units of *a*/60, *b*/30 and *c*/120 using Patterson-

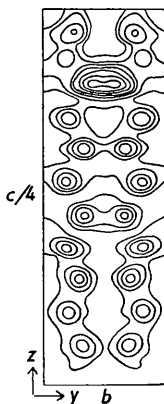


Fig. 3. Projection on (100) for C<sub>9</sub> hydrazide. Contours at arbitrary intervals of electron density.

Tunell strips. The third *F* syntheses on (010) for C<sub>7</sub> and C<sub>8</sub> hydrazides (116 and 98 terms respectively) are shown in Fig. 2, and the first *F* synthesis on (100) for C<sub>9</sub> hydrazide (50 terms) is shown in Fig. 3. Approximate coordinates derived from these projections are listed in Table 2.

Although these results are preliminary and not very accurate, some conclusions can be reached. The structures of these compounds with either an even or odd number of carbon atoms in the chain appear almost the same in (010) projection. There seem to be no consistent irregularities in bond lengths along the chains, nor do the rather short chains appear to deviate from linearity.

The C<sub>8</sub> and C<sub>9</sub> hydrazides have been chosen for further study and three-dimensional integrated data have been collected for the C<sub>9</sub> compound.

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**Anisotropic temperature factors in *hkl* structure-factor calculations.** By D. R. HOLMES, *Imperial Chemical Industries Ltd., Plastics Division, Research Department, Welwyn Garden City, Herts., England*

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In recent crystal-structure analyses it has been found necessary to apply anisotropic temperature factors, either to the structure factors as a whole (Hughes, 1941) or to the structure-factor contributions of particular atoms (Cochran, 1951*a*). Effects closely similar to those of asymmetric thermal vibrations are produced on the observed structure factors of crystals containing atoms whose electron clouds show appreciable departures from spherical symmetry owing to their bonding arrangements. Difference syntheses (Cochran, 1951*b*) enable one to find the direction of maximum thermal vibration (or maximum departure from spherical electron density) and the magnitude of the constants involved, and Cochran himself has shown how to apply anisotropic temperature factors

in two-dimensional structure-factor calculations. The purpose of this note is to show that, by making certain assumptions about the symmetry of the electron-density clouds, such factors may easily be applied to the calculations of general *hkl* structure factors required by accurate crystal-structure analyses. The analysis given applies only to cases in which the vibration ellipsoid of the atom (or electron-density ellipsoid) is an ellipsoid of revolution. The ellipsoid is treated as a linear vibration imposed on a spherically symmetrical vibration, and the effects of both are considered separately. This limitation was imposed so that the final result could be used in a practical case by a simple graphical interpretation.

For the sake of simplicity of explanation the formula